



# Optimization of Particle Size for PLGA Nanoparticles Prepared by the Ouzo Effect

Erik Rytting, Tobias Lebhardt, and Thomas Kissel

*Philipps-Universität Marburg*

# Why Nanoparticles?



- Nanoparticles can be the right tool for the job
- Advantages
  - Enhanced bioavailability and controlled release
  - Penetration across biological barriers
  - Increased drug targeting efficiency

# Methods for Nanoparticle Production



## Materials

- Polymers
- Metals
- Ceramics
- Biological materials
  - Liposomes
  - Peptides
- Polymeric Nanoparticles
  - Nanoparticles formed using **monomers** as the starting point
  - Nanoparticles prepared using **preformed polymers** as the starting point

# Methods Employing Preformed Polymers



- Emulsion Diffusion
  - Interfacial Precipitation
- Emulsion Evaporation
  - Double Emulsion Evaporation
- Salting Out
- Solvent Displacement
  - Nanoprecipitation, Solvent Diffusion, Spontaneous Precipitation, Ouzo Effect

# Solvent Displacement

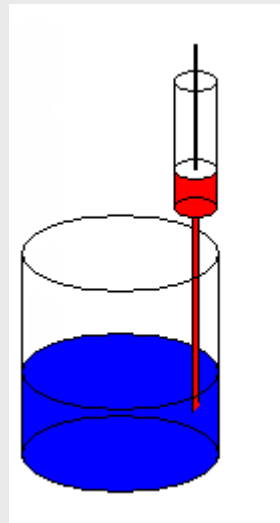


- Polymer and drug are dissolved in a water-miscible organic solvent

# Solvent Displacement



- Polymer and drug are dissolved in a water-miscible organic solvent
- This solution is added to an aqueous solution



# Solvent Displacement



- Polymer and drug are dissolved in a water-miscible organic solvent
- This solution is added to an aqueous solution
- Rapid solvent diffusion leads to instantaneous nanoparticle formation

# Solvent Displacement



- Polymer and drug are dissolved in a water-miscible organic solvent
- This solution is added to an aqueous solution
- Rapid solvent diffusion leads to instantaneous nanoparticle formation
- Solvent is removed by evaporation



# Solvent Displacement



- Advantages
  - Relatively simple process
  - Nontoxic solvents can be used
  - Does not require high shear stress
  - Does not require surfactants or stabilizers
- Disadvantages
  - Solvent evaporation can be time consuming
  - Poor encapsulation of hydrophilic drugs

# Nanoparticle Characterization



- Size
- Surface Characteristics
- Encapsulation efficiency
- Drug Release
- Degradation
- Biocompatibility

# Nanoparticle Characterization



- Size
- Surface Characteristics
- Encapsulation efficiency
- Drug Release
- Degradation
- Biocompatibility

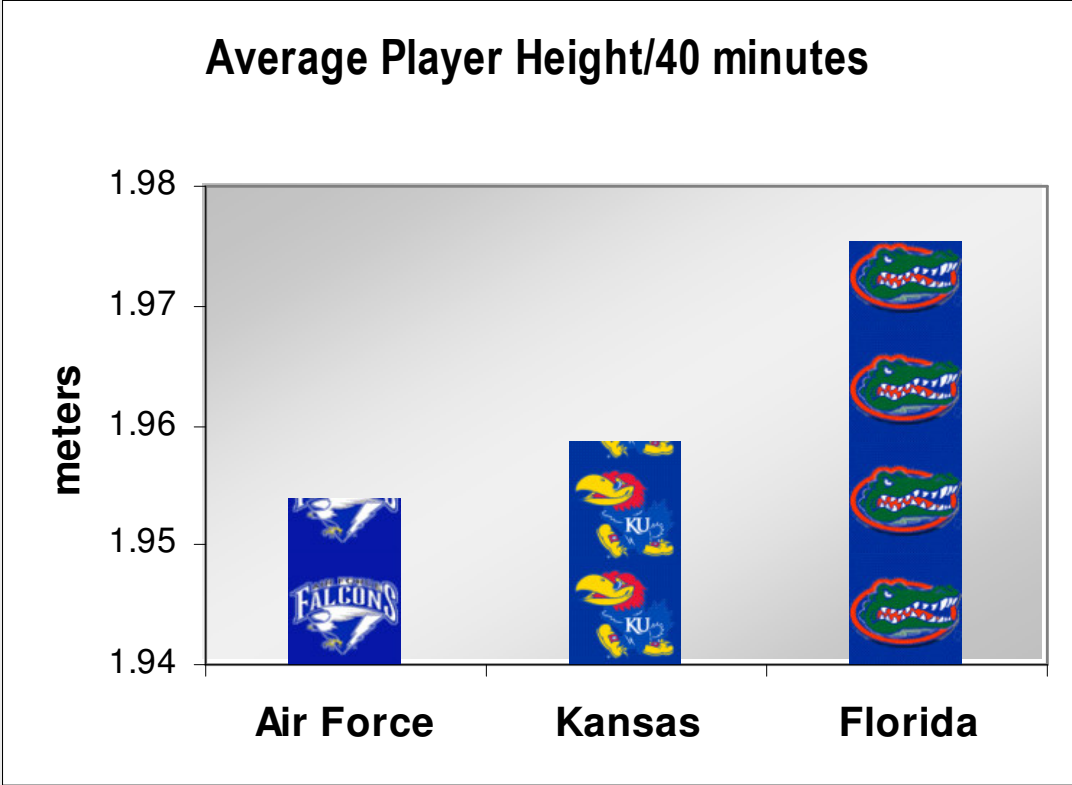
# The Importance of Size



# The Importance of Size



- Basketball



*2005-2006 Season*

# The Importance of Size



<b>NCAA Div. I Rank</b>	<b>Air Force</b>	<b>Kansas</b>	<b>Florida</b>
Rebound Margin	296/326	20/326	59/326
Blocked Shots	281/326	20/326	22/326

# The Importance of Size



- Nanoparticle Size
  - Endothelial Permeability
    - Particle escape from vasculature
      - Liver sinus endothelium (up to 150 nm)
      - Tumor capillaries (up to 300 nm)
    - Blood-brain barrier
  - Splenic Filtration
  - Macrophage Clearance
  - Cellular Uptake
  - Particle Degradation and Drug Release Rates

# Ouzo Effect

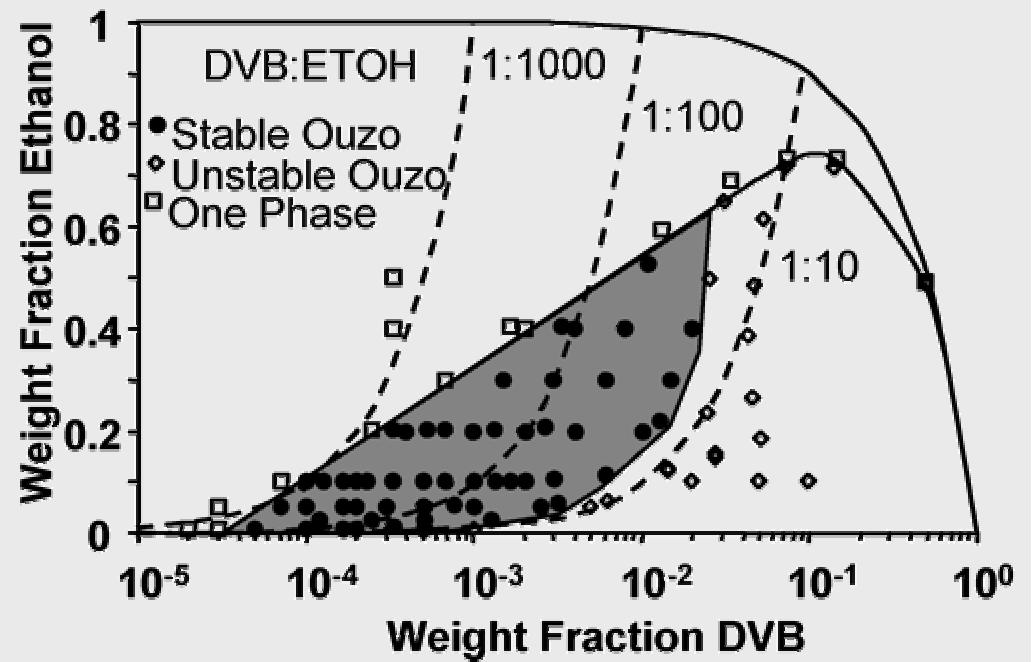
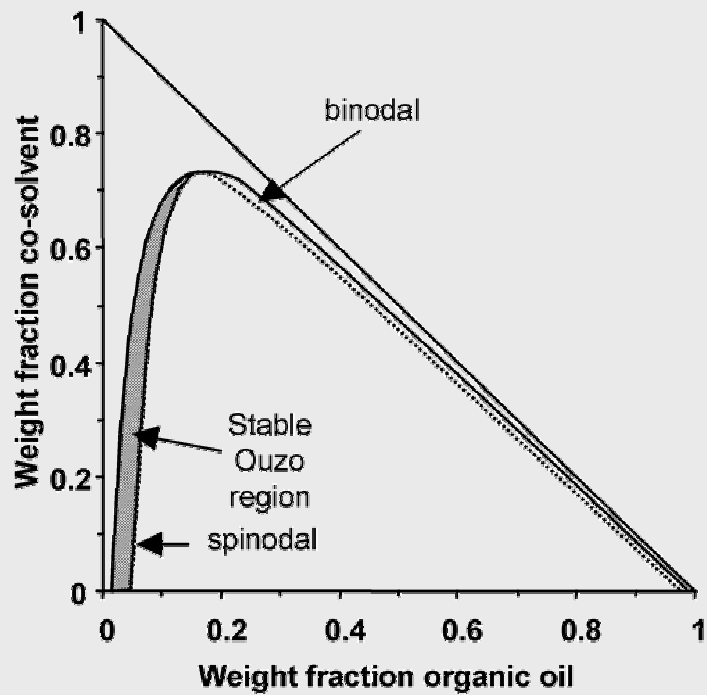


- Solvent displacement is also referred to as the Ouzo effect
  - Ouzo is a mixture of anethol, ethanol, and water (from anis seeds)
  - Additional water leads to micro/nanoparticle formation
  - Solution proportions fall into a thermodynamically metastable region
    - Homogeneous nucleation occurs
    - Leads to a uniform dispersion (milliseconds) followed by Ostwald ripening (seconds)



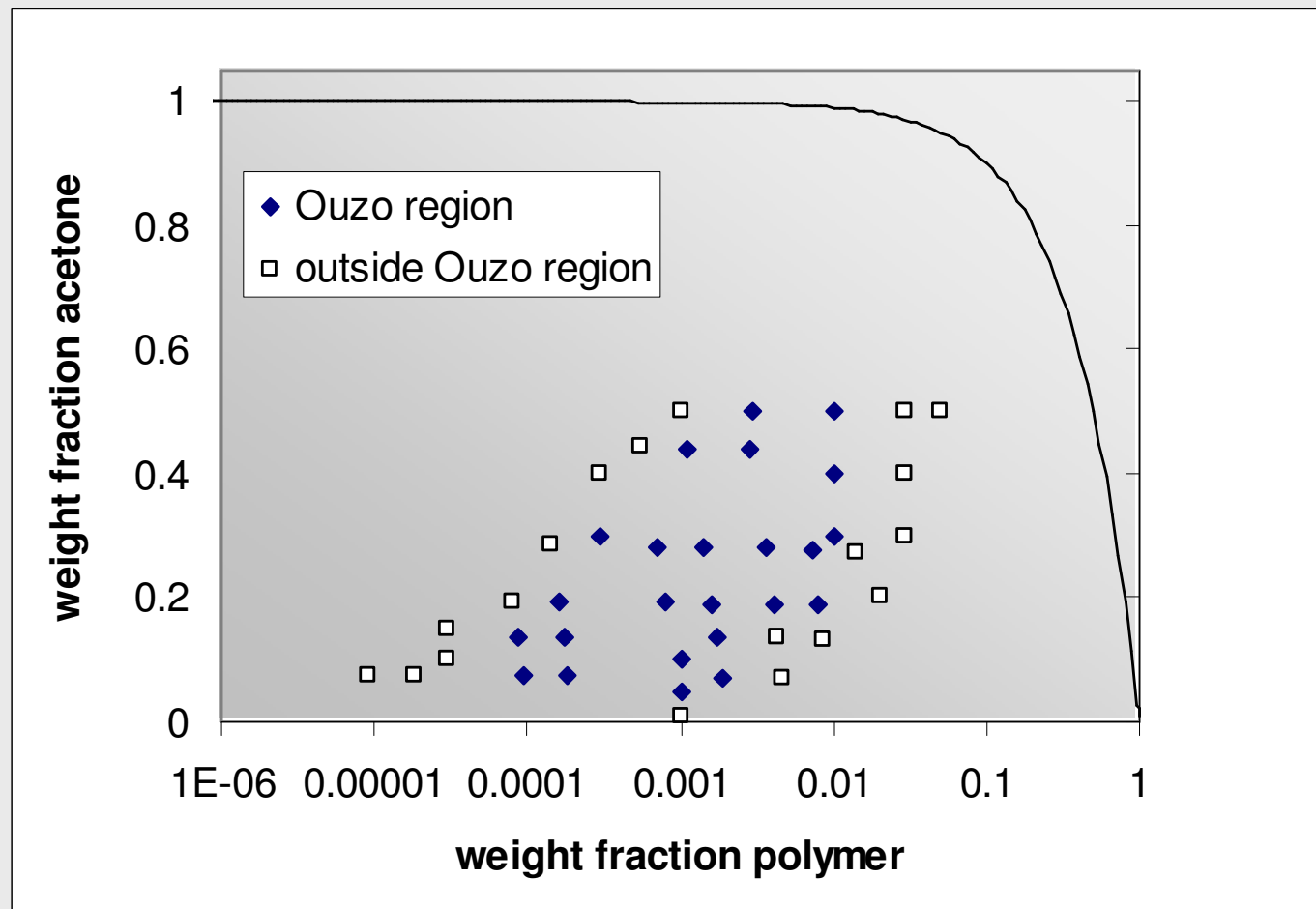


# Ouzo Region

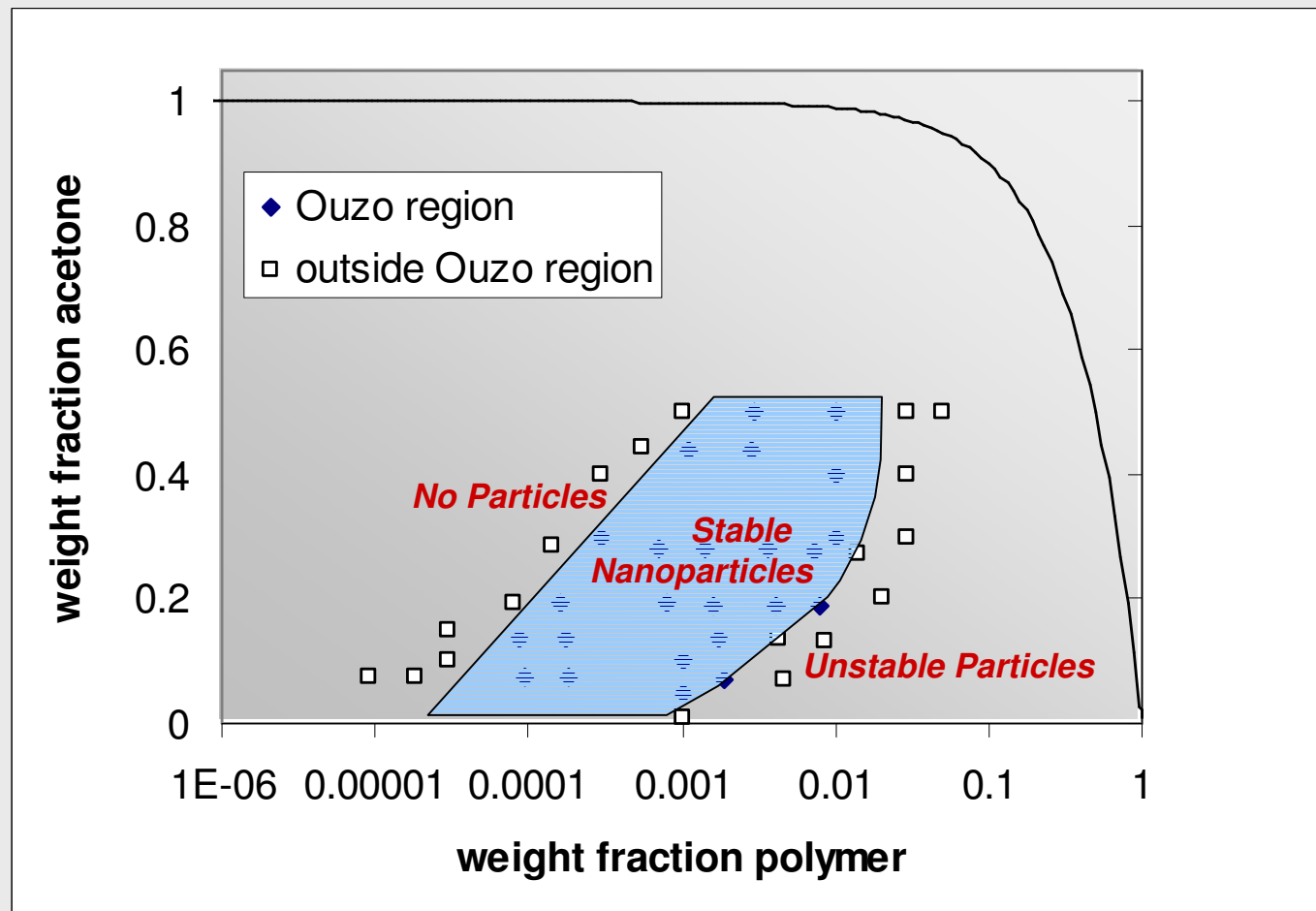


From Vitale and Katz, *Langmuir* 2003, 19, 4105-4110.

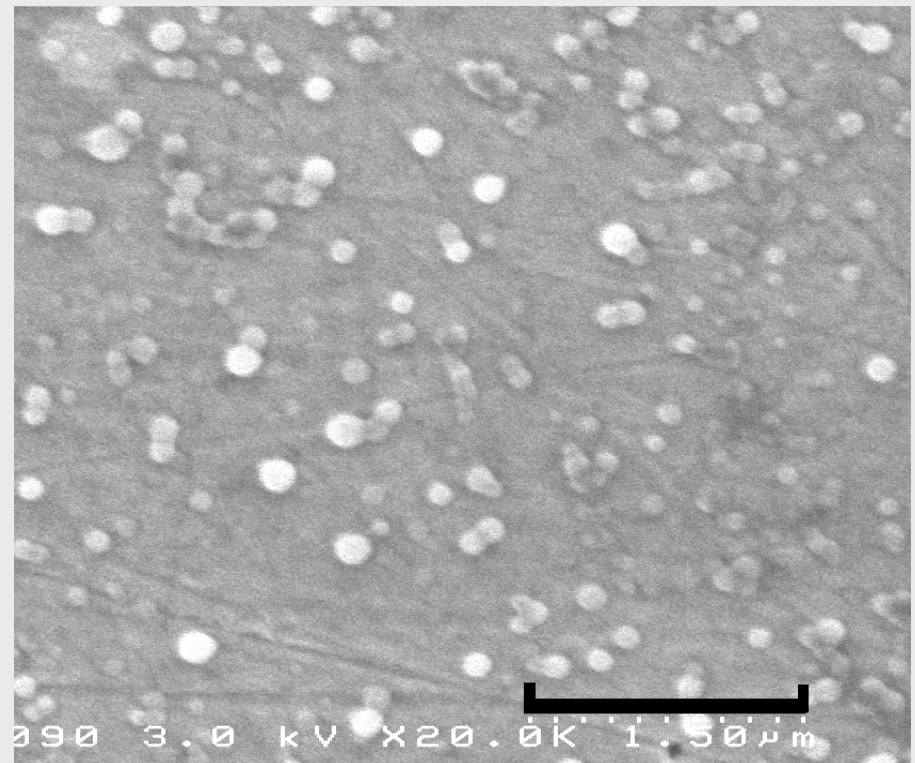
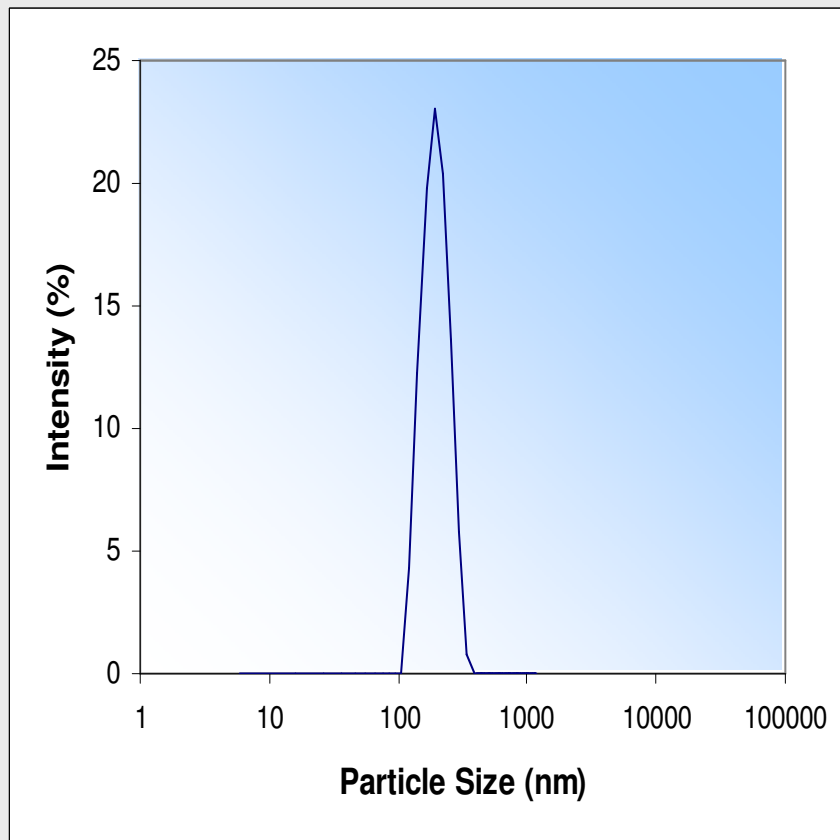
# Ouzo Region for RG502H/Acetone



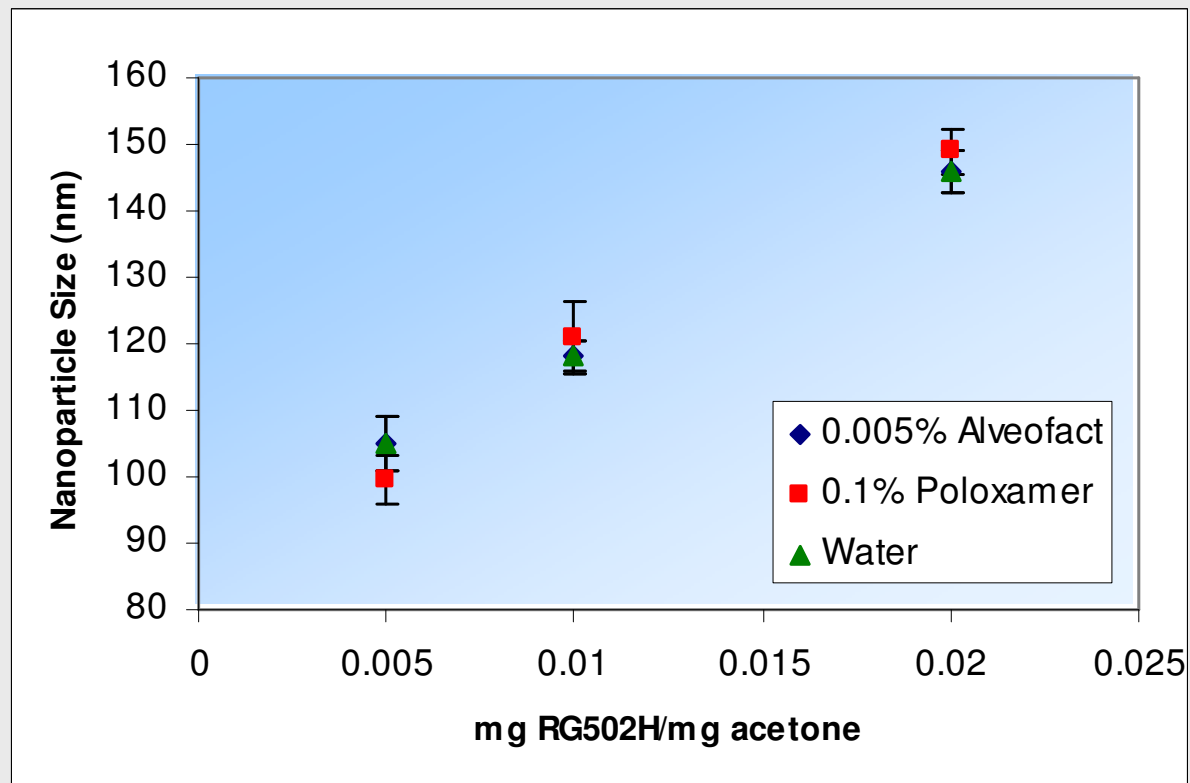
# Ouzo Region for RG502H/Acetone



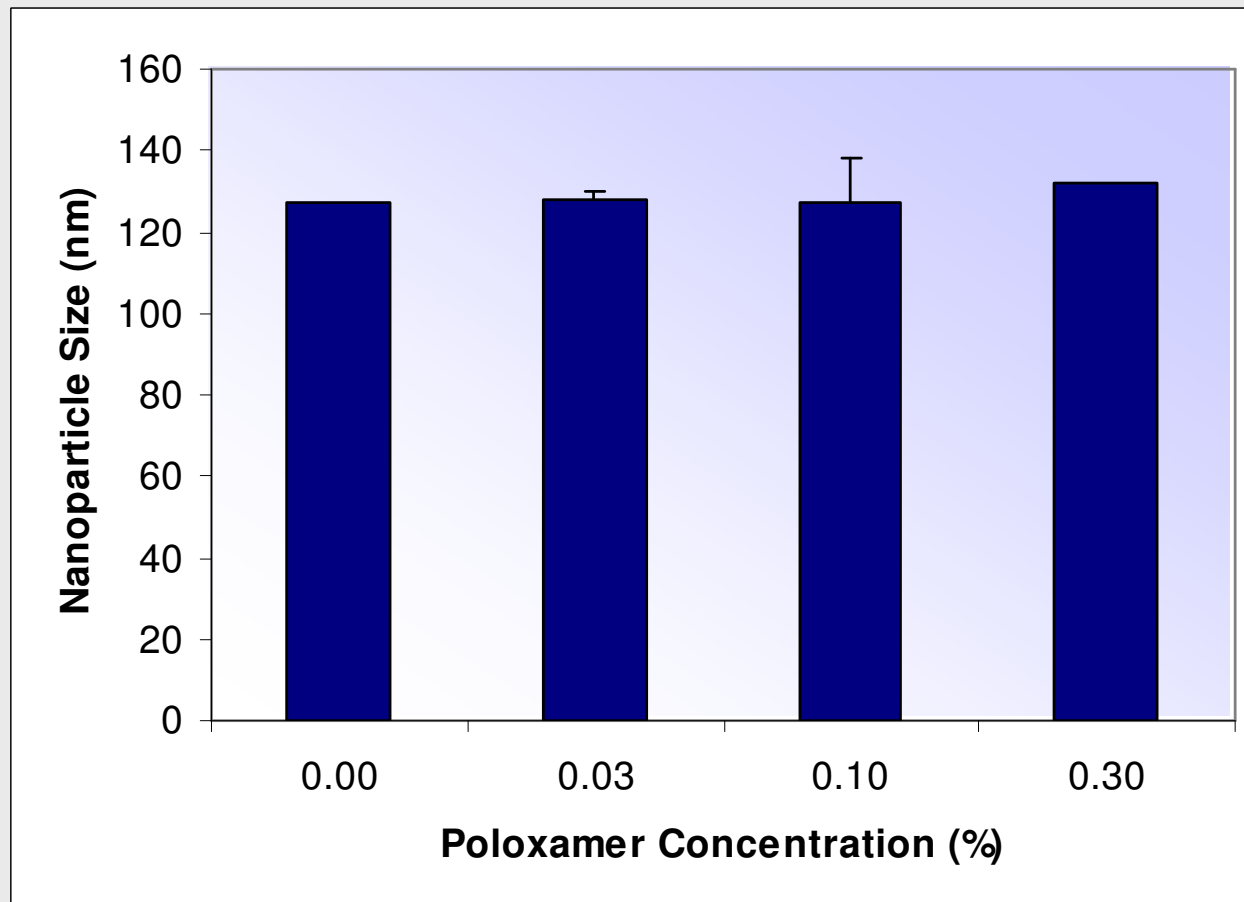
# Particle Size Distribution



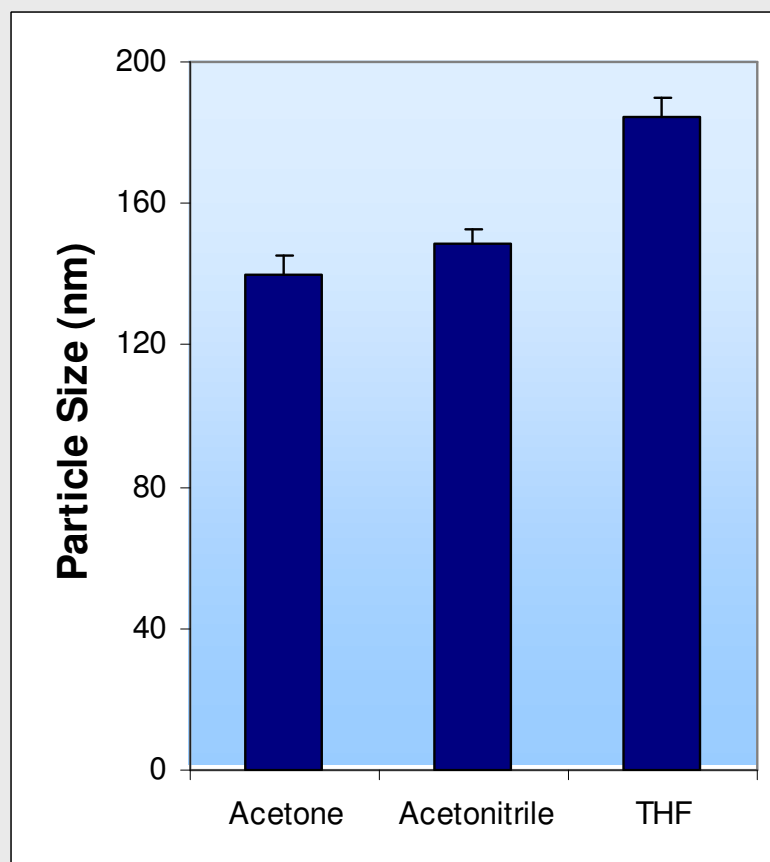
# Surfactants & Polymer Concentration



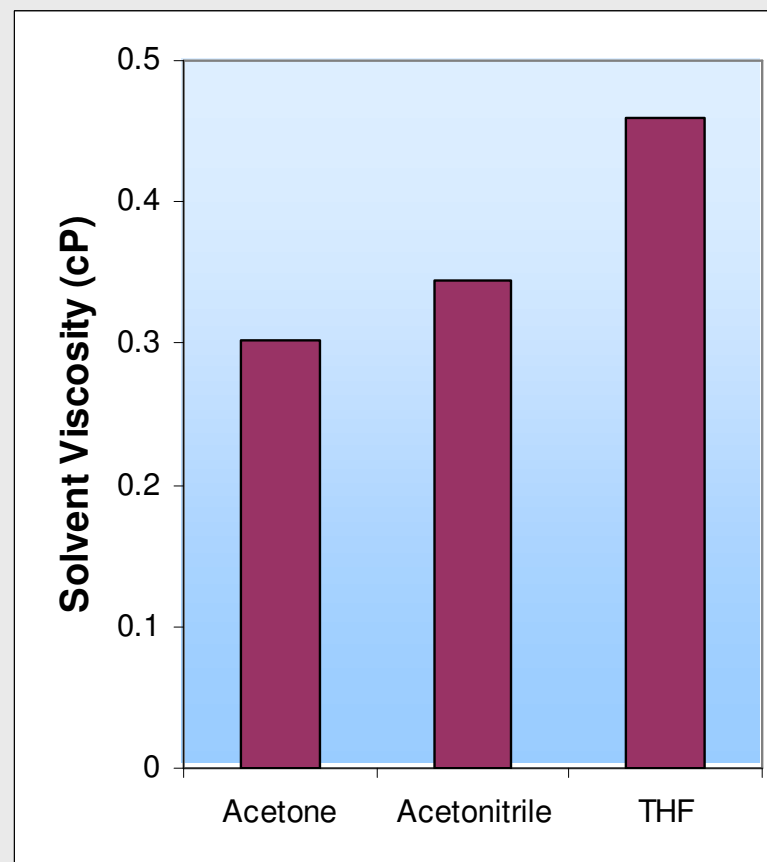
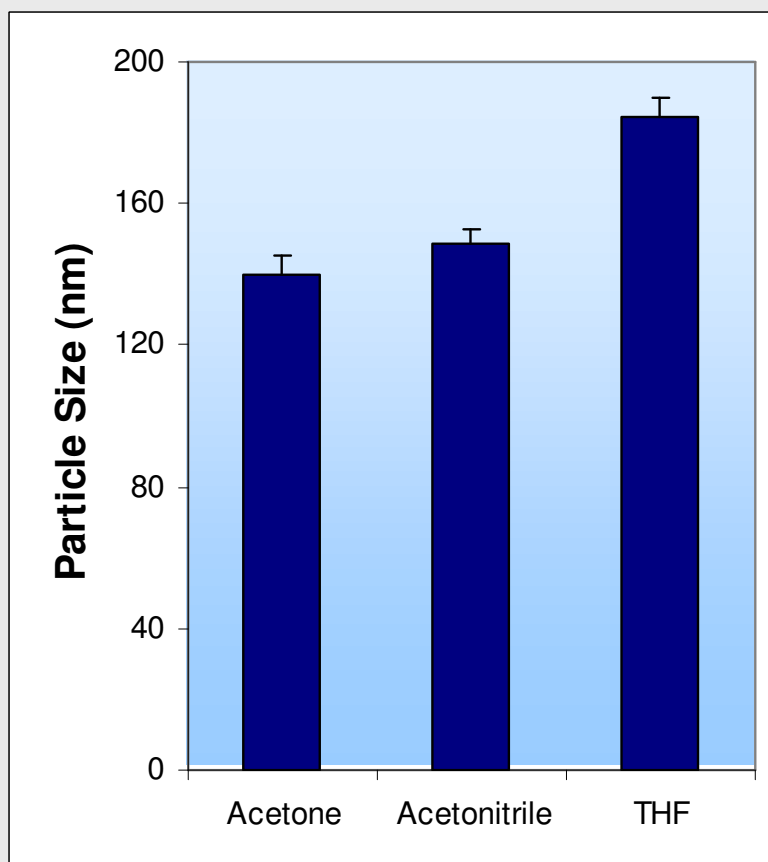
# Surfactant Concentrations



# Different Solvents



# Different Solvents

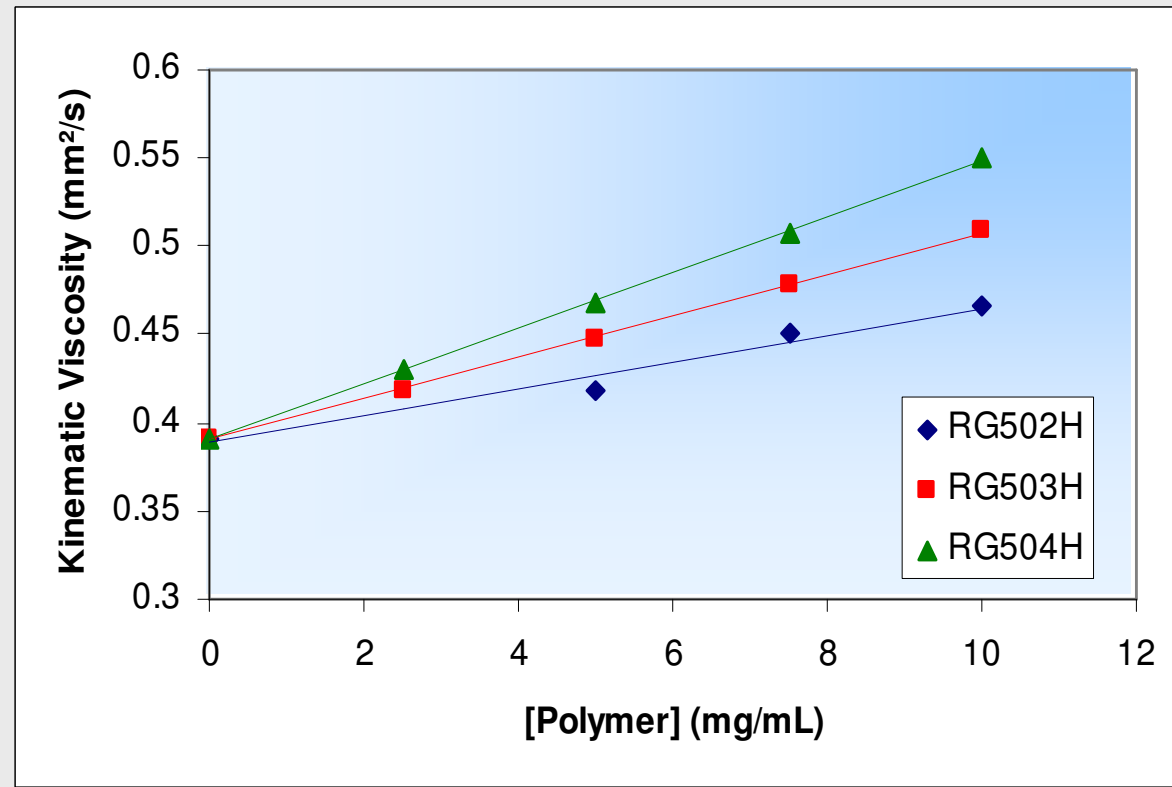




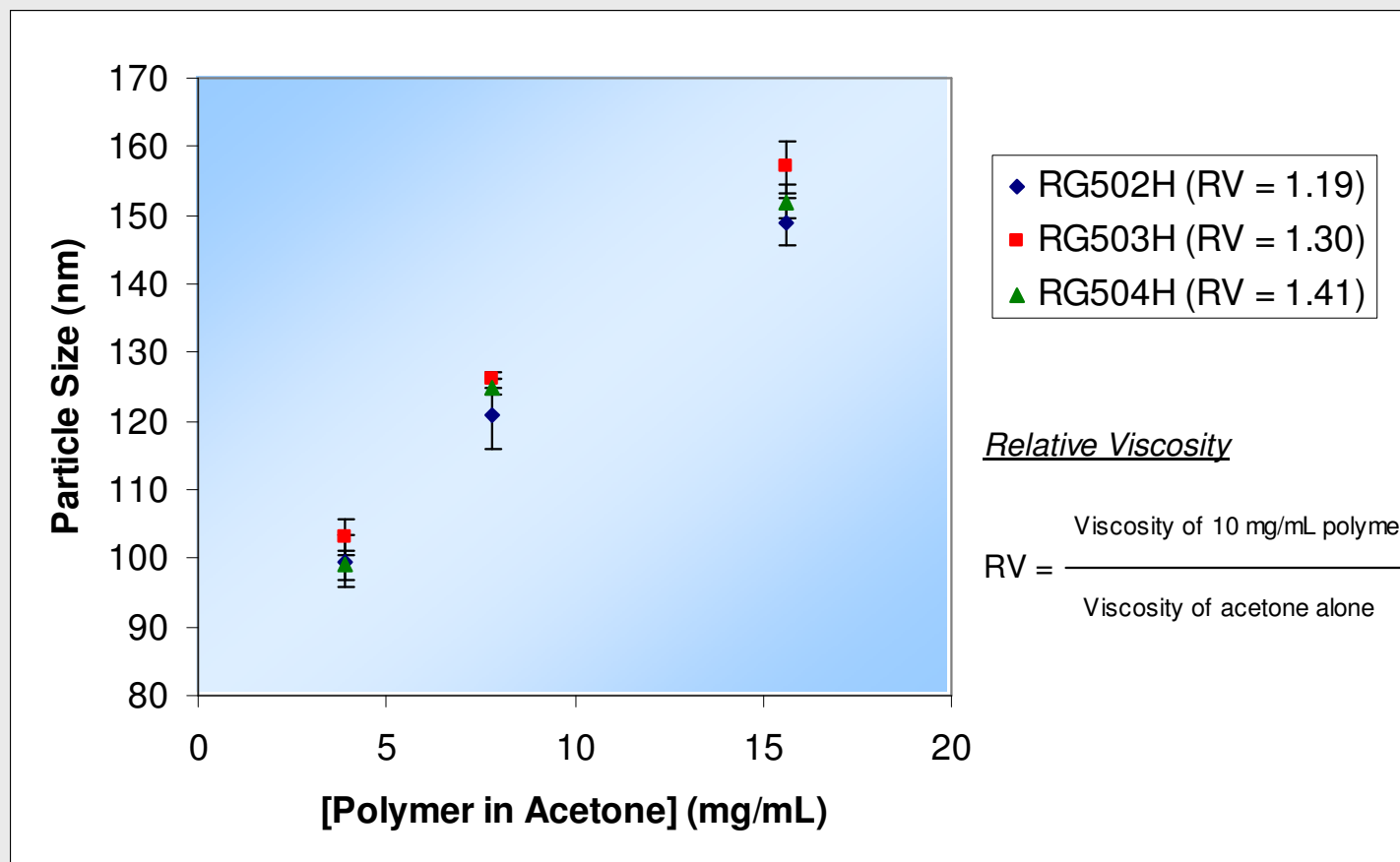
# PLGAs of Increasing MW



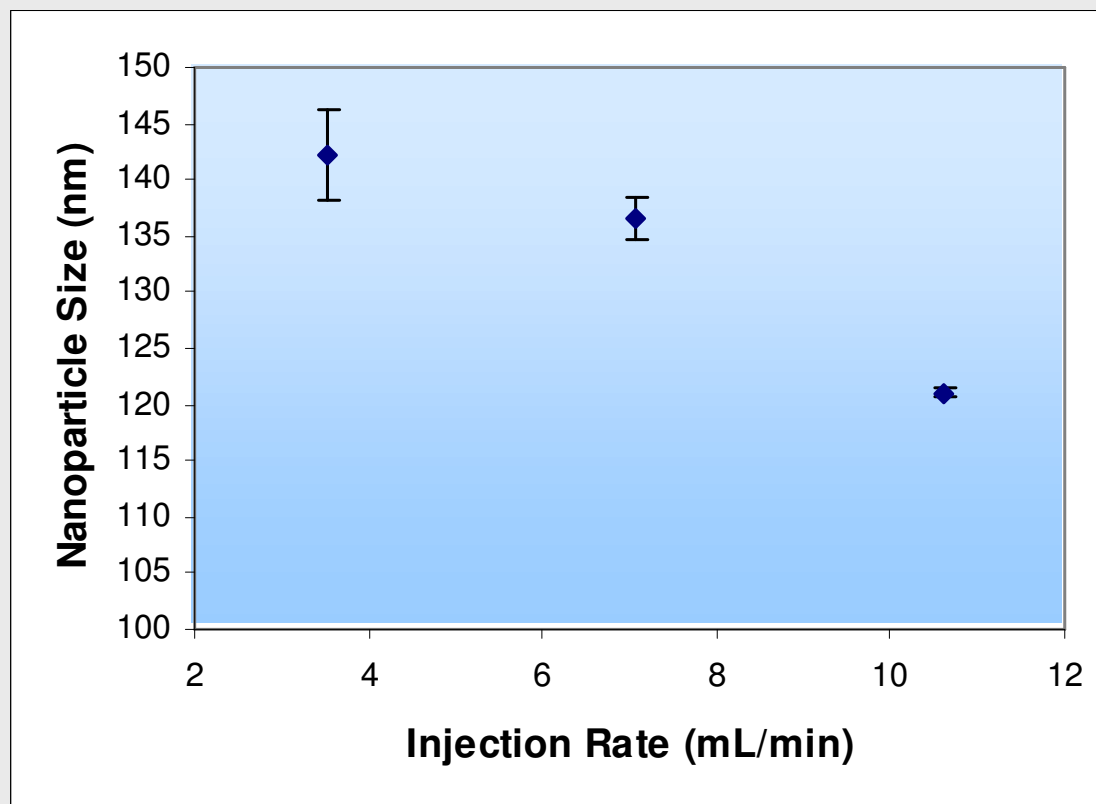
<u>Polymer</u>	<u>M<sub>n</sub> (kDa)</u>
RG502H	12
RG503H	23
RG504H	48



# Viscosity

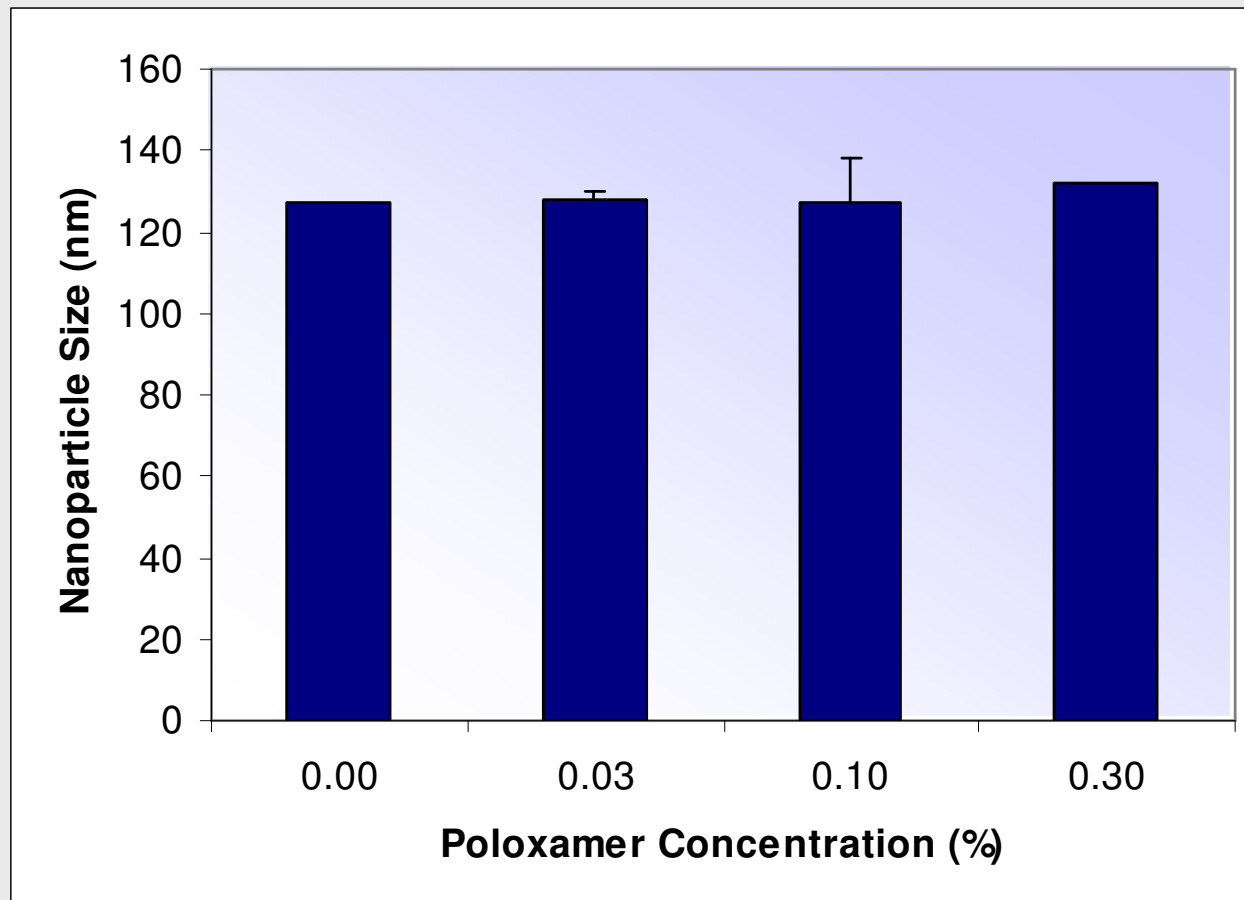


# Injection Rate

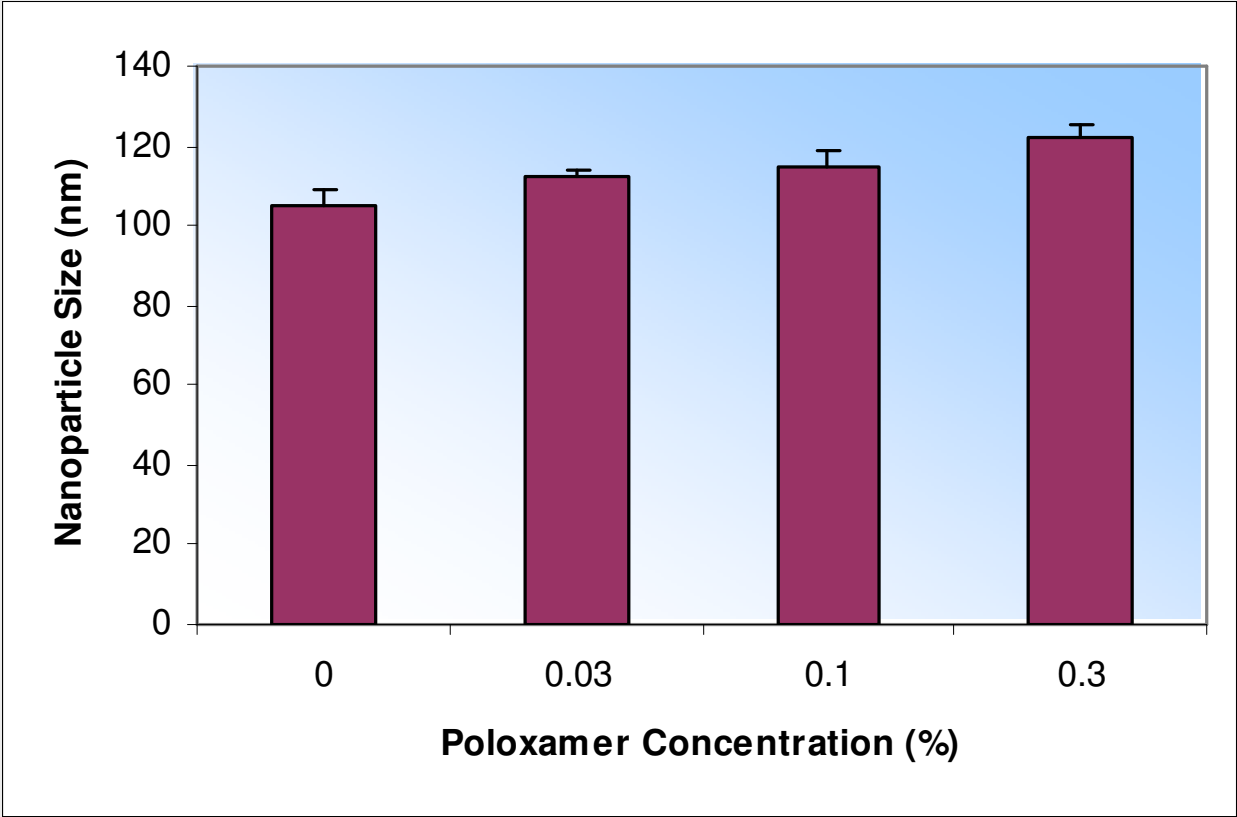


4.0 mg/mL RG 502 H in acetone added to 0.1% Poloxamer.

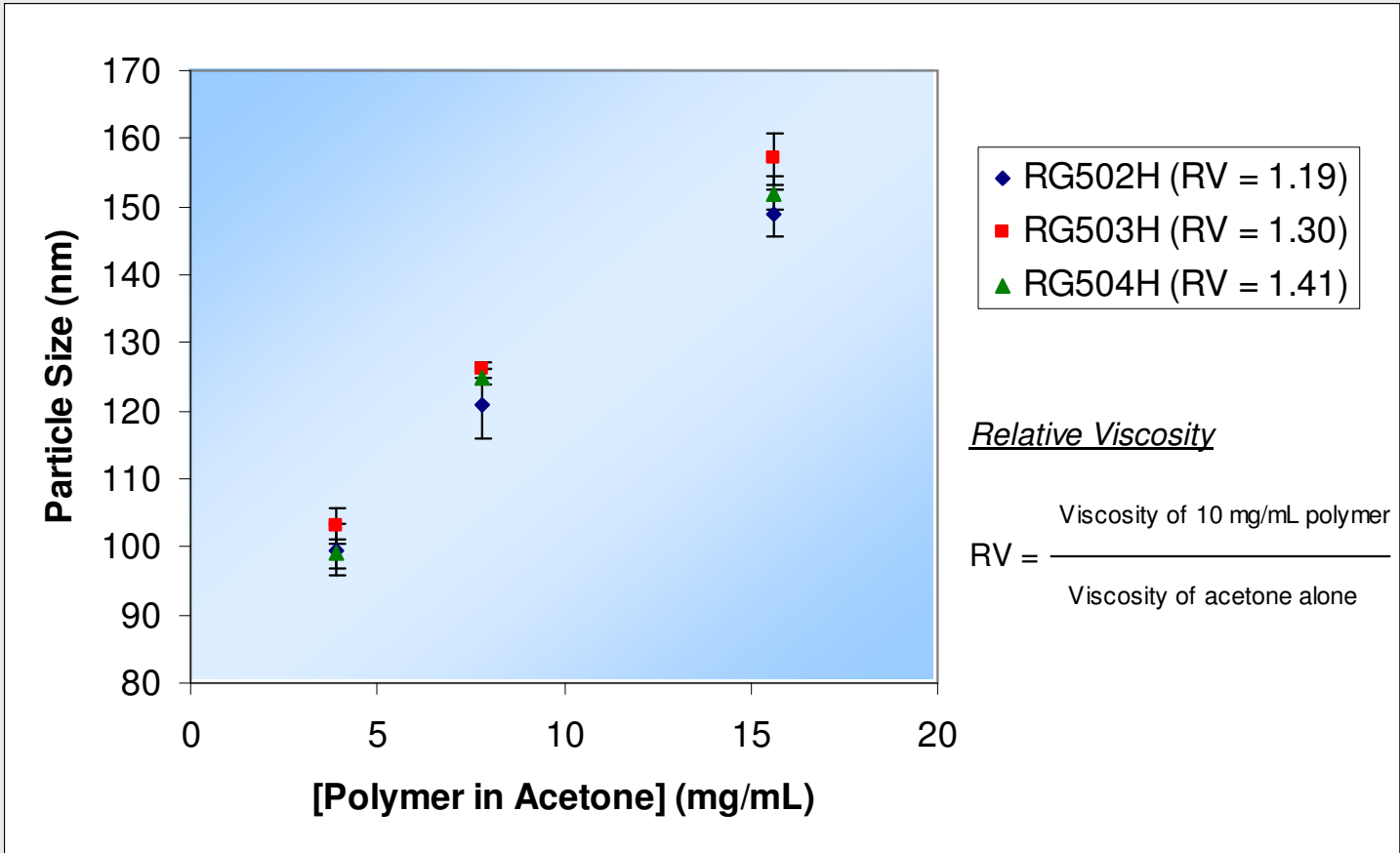
# Surfactants – Fast Injection



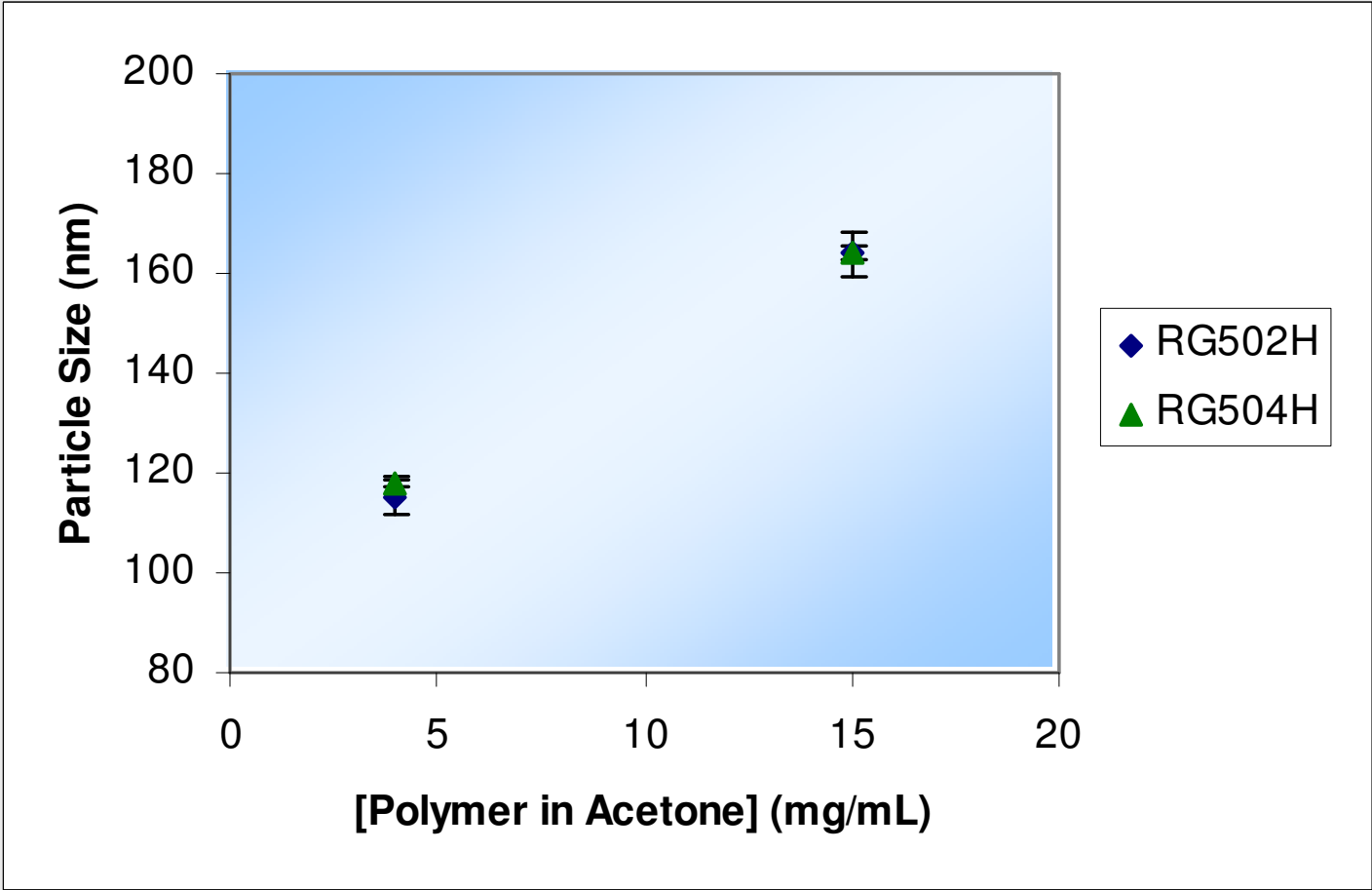
# Surfactants – Slow Injection



# Viscosity – Fast Injection



# Viscosity – Slow Injection



# Importance of Size



	Air Force	Kansas	Florida
Win-Loss Record	24-7	25-8	33-6



# Importance of Size



	Air Force	Kansas	Florida
Win-Loss Record	24-7	25-8	33-6
Scoring Defense	1/326	33/326	50/326
Field Goal % Defense	132/326	1/326	21/326
Scoring Offense	267/326	42/326	22/326

# Importance of Size



	Air Force	Kansas	Florida
Win-Loss Record	24-7	25-8	33-6
Scoring Defense	1/326	33/326	50/326
Field Goal % Defense	132/326	1/326	21/326
Scoring Offense	267/326	42/326	22/326
Personal Fouls/Game	13/326	140/326	29/326
3-Pointers Made/Game	12/326	173/326	70/326
3-Point Shooting %	4/326	44/326	15/326
Free Throw Shooting %	29/326	226/326	34/326

# Improved Drug Delivery



- Nanoparticle Size

# Improved Drug Delivery



- Nanoparticle Size
- Precision/Accuracy

# Improved Drug Delivery



- Nanoparticle Size
- Precision/Accuracy/Targeting

# Improved Drug Delivery



- Nanoparticle Size
- Precision/Accuracy/Targeting
- Other Strategies

# Summary



# Summary



- How can one control Basketball player size?
  - Nutrition
  - Exercise
  - Genetics



# Summary



- How can one control Nanoparticle size?
  - Solvent viscosity
  - Injection speed
  - Polymer concentration

# Acknowledgments



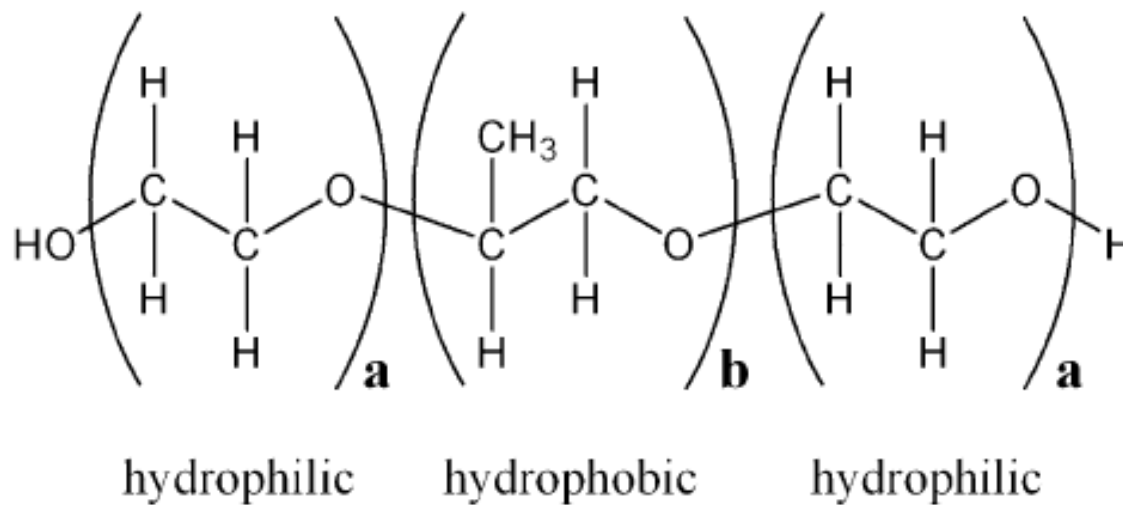
- Co-authors: Tobias Lebhardt & Thomas Kissel
- Others
  - Claudia Packhäuser
  - Theresa Haas
  - Ulrike Nierste

# References



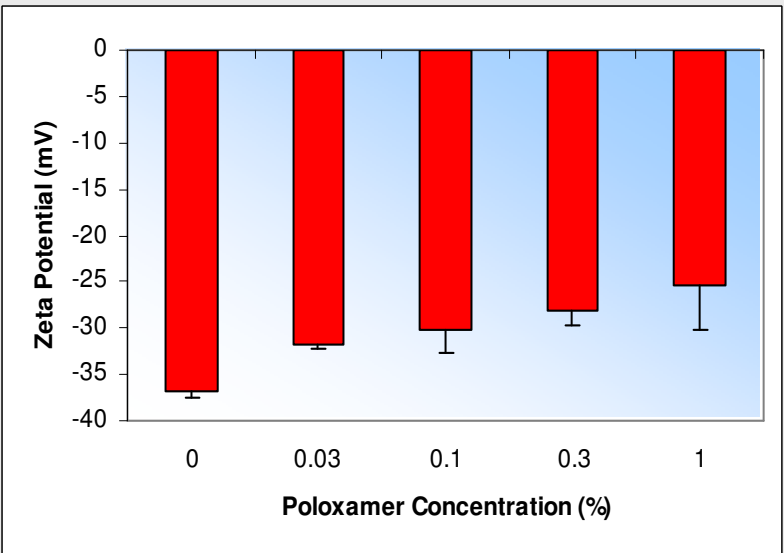
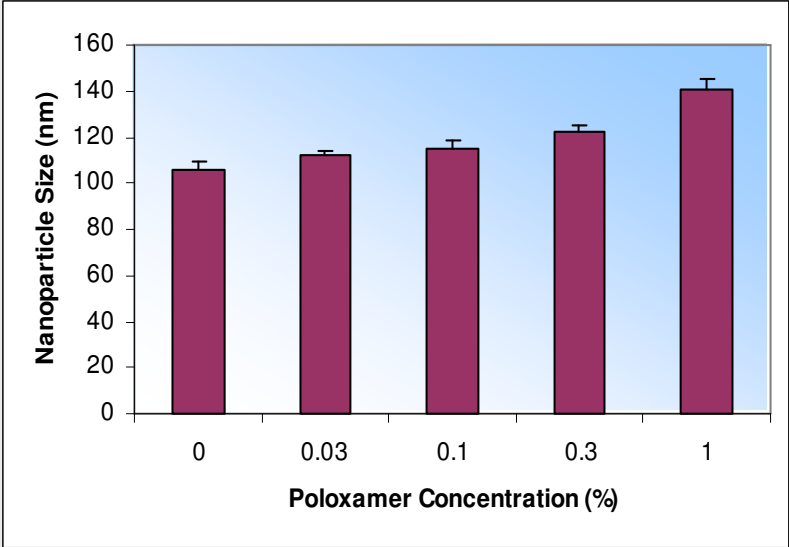
- Astete and Sabliov, *J Biomater Sci Polymer Edn*, 2006, 17, 247-289.
- Avgoustakis, K. *Current Drug Delivery*, 2004, 1, 321-333.
- Bala et al., *Critical Rev Ther Drug Carrier Sys*, 2004, 21, 387-422.
- Desai et al., *Pharm Res*, 1997, 14, 1568-1573.
- Fessi et al., *Int J Pharm*, 1989, 55, R1-R4.
- Ganachaud and Katz, *ChemPhysChem*, 2005, 6, 209-216.
- Gao and Jiang, *Int J Pharm*, 2006, 213-219.
- Hughes, *Nanomedicine*, 2005, 1, 22-30.
- Mayer, *Int J Artif Organs*, 2005, 28, 1163-1171.
- Moghimi et al., *FASEB J*, 2005, 19, 311-330.
- Molpeceres et al., *J Pharm Sci*, 1996, 85, 206-213.
- Panyam and Labhasetwar, *Adv Drug Del Rev*, 2003, 329-347.
- Sahoo and Labhasetwar, *Drug Discovery Today*, 2003, 8, 1112-1120.
- Vitale and Katz, *Langmuir* 2003, 19, 4105-4110.
- Yih and Al-Fandi, *J Cell Biochem*, 2006, 97, 1184-1190.

# Poloxamer 188 NF (Pluronic F68)

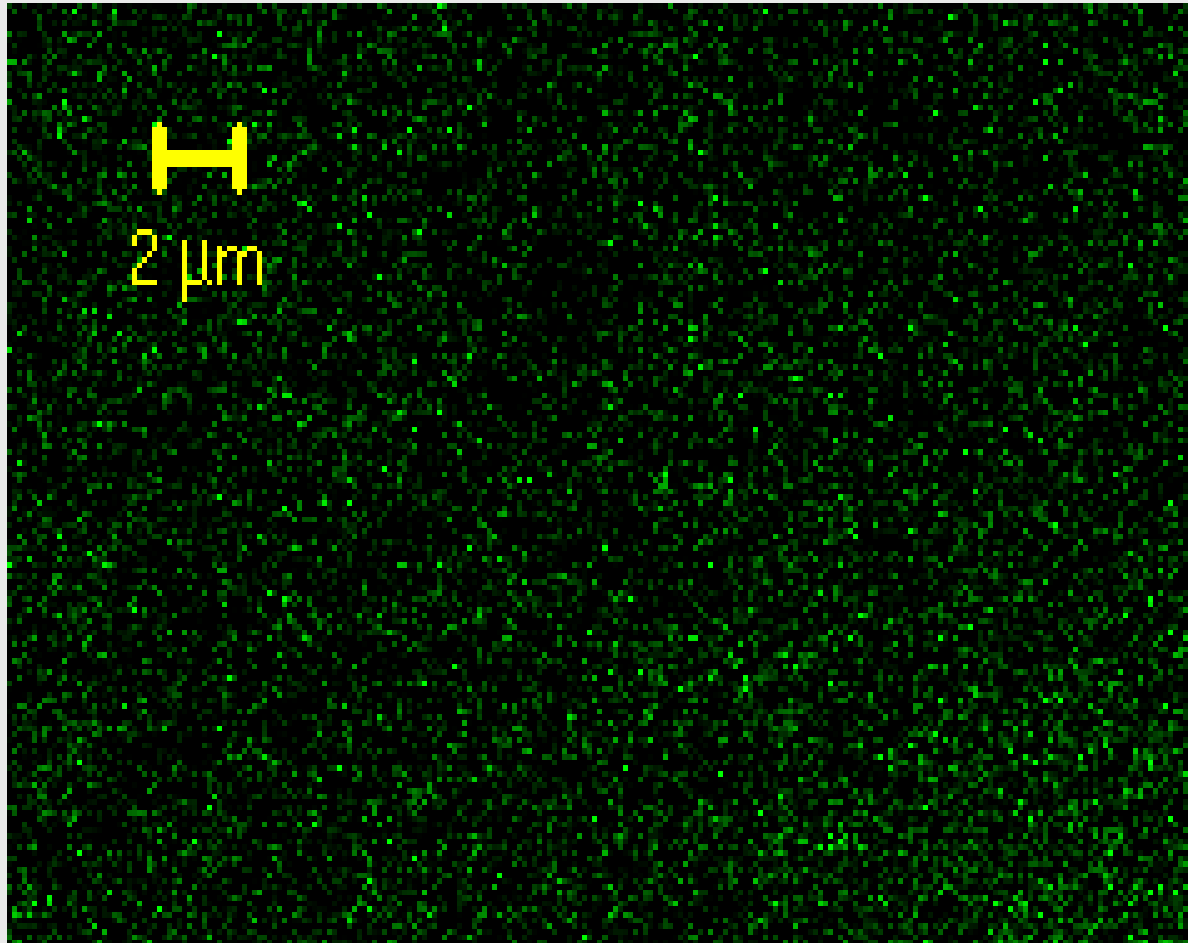


From Walsh et al., *Ann. N.Y. Acad. Sci.* 1066: 321–327 (2005).

# Surface Charge



# Fluorescent Nanoparticles



Nanoparticles prepared with 10 μg Coumarin-6 per 10 mg RG502H in 1 mL acetone,  $Z_{\text{ave}} = 117 \pm 3$  nm. Fluorescence Microscopy settings:  $\lambda_{\text{ex}} = 488$  nm,  $\lambda_{\text{em}} = 515$  nm. Suspension of particles in water after washing in a Sephadex G-50 column.