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# Structure of human muscle creatine kinase

The crystal structure of human muscle creatine kinase has been determined by the molecular-replacement method and refined at 3.5 Å resolution. The structures of both the monomer and the dimer closely resemble those of the other known structures in the creatine kinase family. Two types of dimers, one with a non-crystallographic twofold symmetry axis and the other with a crystallographic twofold symmetry axis, were found to occur simultaneously in the crystal. These dimers form an infinite 'double-helix'-like structure along an unusual long crystallographic  $3_1$  axis.

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PDB Reference: human muscle creatine kinase, 1i0e.

#### 1. Introduction

Creatine kinase (E.C. 2.7.3.2) reversibly catalyzes the phosphoryl transfer between ATP and creatine. It plays an important role in the rapid regeneration of ATP in cells and tissues with high and fluctuating energy demands. Two kinds of creatine kinases (CK) from cytosol and mitochondria have been isolated from many species. Cytosolic CK, a dimer of about 85 kDa, has three isoforms: MM, MB and BB (M and B denote muscle and brain tissues, respectively; Eppenberger et al., 1967). However, mitochondrial CK has two isoforms, ubiquitous Mia-CK and sarcomeric Mib-CK, existing in dimeric and octameric forms (Schlegel, Wyss et al., 1988; Schlegel, Zurbriggen et al., 1988). An elevated level of CK in human blood is an important diagnostic indicator for diseases of the nervous system and the heart muscle, for malignant hypothermia and for certain tumours.

More than 20 creatine kinases have been sequenced (Muhlebach et al., 1994) and they

have high amino-acid sequence identity. Since 1996, several creatine kinase crystal structures have been solved, including mitochondrial CK from sarcomas (sMtCK; Fritz-Wolf et al., 1996), MM-type CK from rabbit (RCK; Rao et al., 1998), BB-type CK from chicken (BB-CK; Eder et al., 1999) and mitochondrial CK from human (uMtCK; Eder, Fritz-Wolf et al., 2000). These structural investigations have provided important information on the folding, active-site conformation and oligomeric states of the enzyme.

The MM-type creatine kinase from human muscle (HCK) was purified and characterized (Yang *et al.*, 1997; Yang & Zhou, 1997) and the gene was cloned, characterized and partially sequenced (Trask *et al.*, 1988). Here, we describe the crystal structure of HCK in the unliganded state at 3.5 Å resolution.

## 2. Materials and methods

The crystallization of and data collection from HCK were reported by Tang  $et\ al.$  (1998). The data were reprocessed with the HKL suite of programs (Otwinowski & Minor, 1997). The space group was reassigned as  $P3_112$  or  $P3_212$ . The crystal has an unusual long cell dimension in the c direction (402.9 Å). The crystallographic asymmetric unit contains three or four monomers, corresponding to  $V_{\rm M}$  values of 3.5 or 2.6 Å  $^3$  Da $^{-1}$  and solvent contents of 64.7 or 53.0%, respectively (Matthews, 1968).

The self-rotation function calculated using the program *POLARRFN* (Collaborative

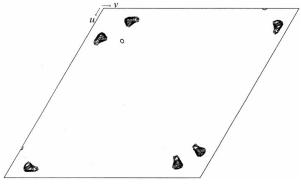


Figure 1 w = 0.25 section of the Patterson map calculated from the data set in the 10.0–4.0 Å resolution range. Contouring starts at the  $4\sigma$  level; the interval is  $1.5\sigma$ .

© 2001 International Union of Crystallography Printed in Denmark – all rights reserved Computational Project, Number 4, 1994) did not yield any significant peaks. However, the Patterson map calculated in the 10.0–4.0 Å resolution range revealed a prominent peak with a height of 6.6% of the origin in the section w=0.249 (Fig. 1). This result suggests that some or even all the monomers in the asymmetric unit had similar orientations and were separated by approximately 1/4 of the cell length along c.

The structure was solved by the molecular-replacement method (Rossmann & Blow, 1962) with the program *AMoRe* (Navaza, 1994). The structure of rabbit muscle creatine kinase determined at 2.4 Å

(Rao et al., 1998) with high sequence identity ( $\sim$ 96%) was selected as a search model. Residues with discrepant sequences were replaced with alanine. In the early stage of the molecular replacement, an angular step of  $2^{\circ}$  and data in the 8.0–4.0 Å resolution range were chosen to calculate the crossrotation function. The results revealed only one significant peak. Three translation-function solutions (R, S, T) were then found from the same cross-rotation function peak. These three monomers had similar orientations. After rigid-body fitting, the model gave a correlation coefficient of 48.7% and an R factor of 45.3%. The electron-density

(a) (b)

**Figure 2** Omit maps for (a) fragment 148–154 located at the dimer interface and (b) the conserved negatively charged cluster (Glu231, Glu2232 and Asp233) which is essential for catalytic activity. The contour level is  $3\sigma$ .

map calculated on the basis of this model appeared to be reasonable. However, structural refinement using the X-PLOR program could not reduce the free R factor below 42%. Careful inspection of the crystal packing revealed that monomer R formed a dimer with the crystallographic twofold symmetry-related monomer R', as did monomer S with S'. These dimers were similar to those of other known structures in the creatine kinase family (Fritz-Wolf et al., 1996; Rao et al., 1998; Eder et al., 1999). However, monomer T had no counterpart, suggesting that one monomer in the asymmetric unit may be lost. A hypothetical monomer, denoted by U, was then constructed so that U could form a dimer with T similar to the RR' or SS' dimers. Further analysis showed that U packs well into the unit cell without close contacts and also adopts an orientation similar to the other monomers.

An attempt to find the fourth monomer by molecular replacement succeeded on using a finer angular step  $(1.2^{\circ})$  within the same resolution range. Under these conditions, a cross rotation-function calculation revealed one significant peak with a correlation coefficient of 16.8% (the maximum noise peak had a correlation coefficient of 8.5%) and the subsequent translationfunction calculation based on the peak yielded four solutions corresponding to the positions of all four monomers (denoted R, S, T, U) in the asymmetric unit. The resulting model had a correlation coefficient of 58.8% and an R factor of 38.2% in the 8.0-4.0 Å resolution range after rigid-body fitting and was used for further refinement. The molecular replacement is summarized in Table 1. The space group was assigned as

The structural refinement was carried out with data in the resolution range 8.0-3.5 Å without a cutoff using the program CNS (Brunger et al., 1998). Model building was performed on an SGI workstation with the program TURBO-FRODO (Roussel & Cambillau, 1989). The refinement was carried out by a combination of positional refinement, simulated annealing and group B-factor refinement. In view of the low data resolution and the multiple copies of the monomers, non-crystallographic symmetry restraints were applied to each monomer (the positional weight constant and  $\sigma$  for the B factor were  $450 \text{ kcal mol}^{-1} \text{ Å}^{-2}$  and  $2.00 \, \text{Å}^2$  for the main-chain atoms and  $400 \text{ kcal mol}^{-1} \text{ Å}^{-2} \text{ and } 3.00 \text{ Å}^{2} \text{ for the side-}$ chain atoms;  $1 \text{ kcal mol}^{-1} \text{ Å}^{-2}$  is equal to  $4.184 \text{ kJ mol}^{-1} \text{ Å}^{-2}$ ). During the refinement, the amino-acid sequence was gradually

changed to that of human creatine kinase. The final round of refinement converged to an R factor of 0.211 and a free R factor of 0.286 ( $R_{\rm free}$  was calculated for 5% of randomly chosen reflections that were excluded from the refinement). A summary of the refinement statistics is shown in Table 2.<sup>1</sup>

#### 3. Results and discussion

## 3.1. Model quality

The final model contains 11 608 non-H protein atoms, with standard deviations from stereochemistry of 0.008 Å for bond lengths and 1.34° for bond angles. In the Ramachandran plot, 76.4% of the nonglycine and non-proline residues are in the most favored regions, 22.9% are in the additional allowed regions and 0.6% are in the generously allowed regions. No residues are within the disallowed region.

In general, most residues have well defined corresponding electron densities in the final electron-density maps calculated with  $2F_o - F_c$  coefficients (Fig. 2). Poor electron densities were observed for the N-terminal segment (residues 1-7) and the C-terminal surface loop (residues 321–331), so these were not included in the final model. Weak electron densities were also found for loops 65-70 and 183-205. It was reported that the latter three loops were all involved in rearrangements of the structure to shield the active site from water during the catalytic reaction procedure (Kabsch & Fritz-Wolf, 1997) and that these disordered loops might undergo substantial backbone movement (up to 15 Å) at the active site during binding of the transition-state analogue (Zhou et al., 1998). Similar structural flexibility for these loops has been observed in other creatine kinases of known structure (Rao et al., 1998; Eder et al., 1999; Eder, Fritz-Wolf et al., 2000).

### 3.2. Crystal packing

The four monomers in the asymmetric unit are denoted A, B, C, D ( $A \leftrightarrow S$ ;  $B \leftrightarrow U$ ;  $D \leftrightarrow T$ ;  $C \leftrightarrow R$ ; S, U, R, T: see Table 1). As described above, monomers A and B form a dimer AB with non-crystallographic twofold symmetry and monomer C (or D) forms a similar dimer but with a crystallographic twofold-axis related monomer C' (or D'). The molecular packing can be described as a 'double-helix' structure (Fig. 3). Two infinite

protein helical chains were formed along the crystallographic  $3_1$  axis (the c axis) using the dimer as a building block and wound around each other. One chain uses dimer AB and its crystallographic equivalents as building blocks, while the other uses dimers CC', DD' and their crystallographic equivalents. However, the dominant interdimer interactions involve dimers from different chains, e.g. A-D' and B-C', using similar contact regions involving  $\beta$ -strand 171–175, turn 176–179 and turn 180–183. This region covers a buried surface area of 283  $\mathring{\rm A}^2$ , as calculated using the program SURFACE

(Collaborative Computational Project, Number 4, 1994).

#### 3.3. Monomer and dimer structures

As expected, the monomer and dimer structures of HCK are similar to those of other creatine kinases, especially RCK. Superposition of the HCK and RCK (PDB code 2crk) monomers for 341  $C^{\alpha}$  atoms gives an r.m.s.d. value of 0.63 Å. This calculation does not include the three flexible segments (residues 65–70, 183–205 and 321–331). Significant differences between these two

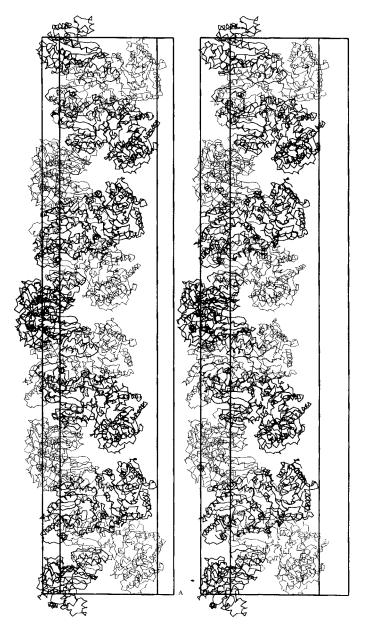


Figure 3 Double-helix-like crystal packing. Two infinite protein helical chains are formed along the crystallographic  $3_1$  axis (the c axis) using the dimer as a building block. Thin lines, using dimer AB and its crystallographic equivalents; thick lines, using dimers CC' and DD' and their crystallographic equivalents.

<sup>&</sup>lt;sup>1</sup> Supplementary data have been deposited in the IUCr electronic archive (Reference: gr2148). Services for accessing these data are described at the back of the journal.

Table 1 Molecular replacement.

Corr, correlation coefficient;  $\alpha$ ,  $\beta$ ,  $\gamma$ , Eulerian angles; x, y, z, fractional Cartesian coordinates. Values in parentheses are the highest noise peaks.

#### Cross-rotation function.

Peak number	α	β	γ	Corr
1	69.78	69.47	100.09	16.8 (8.6)

#### Translation function.

Peak number	x	y	z	Corr	R
1	0.96552	0.26075	0.48009	16.9 (14.0)	52.6 (53.6)

Fixed	Search	х	y	z	Corr	R
1 $1 + 2$ $1 + 2 + 3$	2 3 4	0.16707	0.36529	0.97679	25.9 (17.1) 32.8 (25.6) 36.2 (32.2)	49.1 (50.5)

#### Rigid-body fitting.

α	β	γ	х	y	z	Corr	R	Monomer
71.62	71.68	97.43	0.96666	0.25903	0.48017	58.8	38.9	R
68.71	72.66	102.87	0.06311	0.18529	0.72778	58.8	38.9	T
70.95	67.82	100.21	0.16949	0.36854	0.97677	58.8	38.9	S
72.99	66.29	95.53	0.09650	0.45512	0.22826	58.8	38.9	U

MM-type creatine kinases occur in three surface loops (63–72, 115–122 and 177–205). Because of the high homology of HCK to RCK, this result reflects the conformational flexibility of these loops. The superposition of the HCK and uMtCK (PDB code 1qk1) monomers for 313  $C^{\alpha}$  atoms without including the four obviously different regions (residues 8–14, 177–212, 319–329 and 368–381) gives an r.m.s.d. value of

1.03 Å. The largest differences between these two human-source creatine kinases are found in the N-terminus (residues 1-10 for uMtCK and residues 8-15 for HCK), a surface region (residues 171-203 for uMtCK and residues 176-208 for HCK) and the C-terminus (residues 367-379 for uMtCK and residues 372-381 for HCK). The N-terminus is most likely to be crucial for the octamerization of the Mt enzyme (Kaldis et al., 1994; Eder, Fritz-Wolf et al., 2000). The surface region undergoes a large movement during catalysis in all creatine kinases, as mentioned above. The C-terminus of the Mt enzyme may be related to mitochondrial membrane insertion (Schlattner et al., 1998; Eder, Fritz-Wolf et al., 2000).

The nucleotide-binding site consisting of the conserved positively charged arginine cluster (Arg96, 130, 132, 236, 292, 320 and 341) has a poorly defined

electron density and higher *B* factor, indicating considerable side-chain flexibility, which can be explained by the absence of ATP or its analogues. The conserved negatively charged cluster (Glu231, Glu232 and Asp233) essential for catalytic activity as shown by site-directed mutagenesis experiments (Eder, Stolz *et al.*, 2000) has a well defined electron density (Fig. 2*b*) and shows an arrangement similar to RCK.

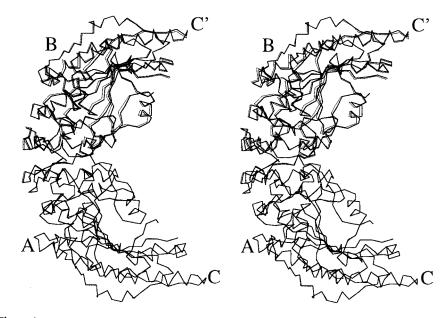


Figure 4
Comparison of AB- and CC'-type dimers after superposition of A and C.

 Table 2

 Data collection and structural refinement statistics.

Data collection	
Space group	$P3_{1}12$
Unit-cell parameters (Å)	a = b = 89.3,
	c = 402.9
Number of monomers per asym. unit	4
Resolution range (Å)	30.0-3.5
No. of observations	63933
No. of unique reflections	21262
Completeness (outermost shell) (%)	88.3 (93.5)
$R_{ m merge}$	0.118
Mosaicity	0.35
Structural refinement	
Resolution range (Å)	8.0-3.5
$R_{ m work}$	0.211
$R_{ m free}$	0.286
Current protein model	11604 protein
	atoms
R.m.s. deviations from ideal value	
Bond lengths (Å)	0.008
Bond angles (°)	1.3
Dihedral angles (°)	22.6
Improper angles (°)	0.83
Mean B values ( $\mathring{A}^2$ )	
Protein atoms	41.4
Main-chain atoms	37.9
Side-chain atoms	44.9
-	

All CK isoforms are known to form very stable dimers. The disintegration of the dimer into monomers only occurs under drastic conditions along with partial unfolding of the monomer (Gross et al., 1995; Couthon et al., 1997). This has been shown for all published CK crystal structures. In MM-type RCK, the dimer has a twofold crystallographic symmetry axis. However, in chicken mitochondrial Mib-CK, the dimer has a twofold non-crystallographic symmetry axis. Comparison of the two dimers revealed minor differences (Rao et al., 1998). Interestingly, both types of dimers, one with a non-crystallographic symmetry axis (AB) and the other with a crystallographic symmetry axis (CC' or DD'), were found to simultaneously occur in the structure of HCK presented here. The superposition of AB and CC' or DD' gives  $C^{\alpha}$ r.m.s. deviations of 0.25 Å. A 1.43° rotation was required for the superposition of B onto C' after a least-squares superposition of Aonto C (Fig. 4). At the present resolution, it is not clear whether this minor difference implies hinge-bending flexibility of the dimer, probably induced by crystallographic effects, or is just a reflection of the experimental error.

The intermolecular interactions in dimer *CC'* includes 22 hydrogen bonds, two salt bonds (Table 3) and a number of hydrophobic interactions. The residues involved in the intermolecular interactions include Asn8–Lys9, Tyr14, Glu18–Asp22, Lys25, Ser49, Asp54, Asp62, His145–Arg152, Lys177 and Arg209–Asp210. The intermolecular interactions in dimer *AB* are

# short communications

 Table 3

 Intersubunit hydrogen bonds.

Distances greater then 3.5 Å are not given.

Dimer CC   Dimer CC	
Glu19 OE1         Gly149 N         3.2         2.7           Glu19 OE1         Ser147 OG         2.7         2.6           Glu19 OE2         Ser147 OG         3.2         —           Glu19 OE2         Arg148 N         3.3         3.1           Glu19 OE2         Gly149 N         2.7         2.8           Glu19 O         Arg148 NH1         2.8         2.6           Tyr20 O         Arg152 NH1         —         3.0           Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser147 OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE         2.7         2.8	r AB
Glu19 OE1         Ser147 OG         2.7         2.6           Glu19 OE2         Ser147 OG         3.2         —           Glu19 OE2         Arg148 N         3.3         3.1           Glu19 OE2         Gly149 N         2.7         2.8           Glu19 O         Arg14 NH18         2.8         2.6           Tyr20 O         Arg152 NH1         —         3.0           Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser147 OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE         2.7         2.8	
Glu19 OE2         Ser147 OG         3.2         —           Glu19 OE2         Arg148 N         3.3         3.1           Glu19 OE2         Gly149 N         2.7         2.8           Glu19 O         Arg14 NH18         2.8         2.6           Tyr20 O         Arg152 NH1         —         3.0           Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser14 7OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE         2.7         2.8	
Glu19 OE2         Arg148 N         3.3         3.1           Glu19 OE2         Gly149 N         2.7         2.8           Glu19 O         Arg14 NH18         2.8         2.6           Tyr20 O         Arg152 NH1         —         3.0           Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser14 7OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.1           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE         2.7         2.8	
Glu19 OE2         Gly149 N         2.7         2.8           Glu19 O         Arg14 NH18         2.8         2.6           Tyr20 O         Arg152 NH1         —         3.0           Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser14 TOG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.1           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE2         2.7         2.8	
Glu19 O         Arg14 NH18         2.8         2.6           Tyr20 O         Arg152 NH1         —         3.0           Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser147 OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE2         2.7         2.8	
Tyr20 O         Arg152 NH1         —         3.0           Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser14 7 OG         Glu19 OE1         2.7         2.6           Ser14 7 OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.1           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE2         2.7         2.8	
Tyr20 O         Arg152 NH2         2.7         3.3           Asp54 OD1         Arg148 NH1         2.9         2.7           Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser147 OG         Glu19 OE2         2.7         2.6           Ser14 7OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.2         3.           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 OE2         2.7         2.8	
Asp54 OD1     Arg148 NH1     2.9     2.7       Asp54 OD1     Arg148 NH2     —     2.7       Gln58 NE2     Asp210 OD1     3.2     3.3       Asp62 OD1     Asp210 N     2.6     2.8       Arg152 NH1     Glu18 O     2.7     2.9       Arg149 N     Glu19 OE1     3.2     —       Ser147 OG     Glu19 OE1     2.7     2.6       Ser14 7OG     Glu19 OE2     3.2     3.1       Arg148 N     Glu19 OE2     3.3     —       Gly149 N     Glu19 OE2     2.7     2.8       Arg148 NH1     Glu19 OE     2.8     3.0	
Asp54 OD1         Arg148 NH2         —         2.7           Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         —           Ser147 OG         Glu19 OE1         2.7         2.6           Ser14 7OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.3         —           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 O         2.8         3.0	
Gln58 NE2         Asp210 OD1         3.2         3.3           Asp62 OD1         Asp210 N         2.6         2.8           Arg152 NH1         Glu18 O         2.7         2.9           Arg149 N         Glu19 OE1         3.2         -           Ser147 OG         Glu19 OE1         2.7         2.6           Ser14 7OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.3         -           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 O         2.8         3.0	
Asp62 OD1     Asp210 N     2.6     2.8       Arg152 NH1     Glu18 O     2.7     2.9       Arg149 N     Glu19 OE1     3.2     -       Ser147 OG     Glu19 OE1     2.7     2.6       Ser14 7OG     Glu19 OE2     3.2     3.1       Arg148 N     Glu19 OE2     3.3     -       Gly149 N     Glu19 OE2     2.7     2.8       Arg148 NH1     Glu19 O     2.8     3.0	
Arg152 NH1     Glu18 O     2.7     2.9       Arg149 N     Glu19 OE1     3.2     —       Ser147 OG     Glu19 OE1     2.7     2.6       Ser14 7 OG     Glu19 OE2     3.2     3.1       Arg148 N     Glu19 OE2     3.3     —       Gly149 N     Glu19 OE2     2.7     2.8       Arg148 NH1     Glu19 O     2.8     3.0	
Arg149 N     Glu19 OE1     3.2     —       Ser147 OG     Glu19 OE1     2.7     2.6       Ser14 7OG     Glu19 OE2     3.2     3.1       Arg148 N     Glu19 OE2     3.3     —       Gly149 N     Glu19 OE2     2.7     2.8       Arg148 NH1     Glu19 O     2.8     3.0	
Ser147 OG         Glu19 OE1         2.7         2.6           Ser147 OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.3         -           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 O         2.8         3.0	
Ser14 7OG         Glu19 OE2         3.2         3.1           Arg148 N         Glu19 OE2         3.3         -           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 O         2.8         3.0	
Arg148 N         Glu19 OE2         3.3         -           Gly149 N         Glu19 OE2         2.7         2.8           Arg148 NH1         Glu19 O         2.8         3.0	
Gly149 N Glu19 OE2 2.7 2.8 Arg148 NH1 Glu19 O 2.8 3.0	
Arg148 NH1 Glu19 O 2.8 3.0	
Arg152 NH1 Tyr20 O - 2.6	
Arg152 NH2 Tyr20 O 2.7 -	
Arg148 NH1 Asp54 OD1 2.9 2.9	
Asp210 OD1 Gln58 NE2 3.2 3.1	
Asp210 N Asp62 OD1 2.6 2.6	

similar to those in *CC'*, indicating the stability of the dimer interface. Moreover, comparison of RCK with uMtCK revealed that the interactions are also very similar in either the contact regions or the interaction number. In the HCK structure, the surface

area buried by formation of the dimer is  $2192 \text{ Å}^2$  for CC', which is quite similar to the value (2200 Ų) in uMtCK (Eder, Fritz-Wolf *et al.*, 2000) and corresponds to 6.7% of the monomer surface area. All these results support the view that the dimer in creatine kinase is well conserved.

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